

Product Name

Product Type

PVC Resin

RY-S-QA-T003 Safety Data Sheet: PVC Resin

Issue date: 01 October 2025 Revised no. 11

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SECTION 1: IDENTIFICATION OF THE SUBSTANCE/MIXTURE AND OF THE COMPANY/UNDERTAKING

1.1 Product identifier

Product Name POLYVINYL CHLORIDE (PVC)

1.2 Other means of identification None

1.3 Recommended use of the chemical and restrictions on use

Identified Use Raw material in plastic industry.

Restrictions on use No data are available.

1.4 Details of the manufacturer

Company Identification Thai Plastic and Chemicals Public Co Ltd

Company Address Head office:

1 Siam Cement Road, Bangsue, Bangkok 10800

Plant:

8, I-1 RD, MAP TA PHUT INDUSTRIAL ESTATE, MUANG, RAYONG 21150

Telephone: +66 38 925 200 ext. 6183

1.5 Emergency telephone number

Emergency Phone No. +66 38 925 200 ຕ່ວ 6183

SECTION 2: HAZARDS IDENTIFICATION

2.1 Classification of the substance or mixture

GHS Classification, 10th edition Specific target organ toxicity - single exposure, Category 3 (Respiratory tract

(2023) irritation)

2.2 GHS label elements including precautionary statement

Labelling according to GHS 10th edition (2023)

Hazard pictograms



Signal word Warning

Hazard statements H335 May cause respiratory irritation.

Precautionary statements

Prevention P261 Avoid breathing dust.

P271 Use only outdoors or in a well-ventilated area.

Response P304+P340 IF Inhaled: Remove person to fresh air and keep comfortable for

breathing.

P319 Get medical help if you feel unwell.

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Storage P405 Store locked up.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

Disposal P501 Dispose of contents/container to licensed waste disposal.

2.3 Other hazards with do not

result in classification

No data are available.

SECTION 3: COMPOSITION/INFORMATION ON INGREDIENTS

Distinction of Substance or

Substance

Mixture:

Ingredients	CAS No. Concentration (%w	
Polyvinyl Chloride Homopolymer		
Other Name: Polyvinyl Chloride; PVC;		
Chloroethylene Polymer; Ethene,	9002-86-2	>99.5%
chloro-homopolymer		
Chemical formula: (C ₂ H ₃ Cl) _n		

SECTION 4: FIRST AID MEASURES

4.1 Description of necessary first

aid measures

Inhalation IF Inhaled: Remove person to fresh air and keep comfortable for breathing. If

breathing is difficult. Get medical attention.

Skin contact Wash exposed skin thoroughly with soap and water. If irritation occur, get

medical attention.

Eye Contact Wash eye thoroughly with large amount of water for 15 minutes, ensuring

eyelids are held open. Get medical advice if any pain or redness develops or

persists.

Ingestion If a large amount is taken, see a doctor.

4.2 Most important symptoms

and effects, acute and

delayed

Eye contact may cause redness. Inhalation of this substance may cause

coughing and irritation to the throat and lungs. [1][2]

4.3 Indication of immediate

medical attention and special

treatment needed, if

necessary

None



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SECTION 5: FIRE-FIGHTING

5.1 Suitable Extinguishing media Carbon dioxide (CO₂), water spray.

5.2 Special hazards arising from the chemical

Hydrogen chloride carbon monoxide and carbon dioxide. It may decompose into benzene, aromatic, and aliphatic hydrocarbons.

Since PVC contains chlorine, uncontrolled burning or open-air incineration can produce dioxins during the combustion process. [3]

5.3 Specific protective actions for fire-fighters

Evacuate people from the fire area. Isolate the area and designate it as a restricted, hazardous zone. If the situation is not dangerous, move chemical containers away from the fire. Firefighters should use standard protective equipment. In confined spaces, firefighters should use a self-contained breathing apparatus (SCBA).

SECTION 6: ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions,
protective equipment and
emergency procedures

For non-emergency personnel Avoid inhaling dust. Follow your organization's emergency plan.

For emergency personnel Eliminate all ignition sources. Provide adequate ventilation. Stop leak if safe to

do so. Wear protective clothing as described in Section 8 of this safety data

sheet. Avoid inhaling dust.

6.2 Environmental precautions Prevent entry into waterways, sewers and surface water reservoir. Contain

released product to prevent further contamination of soils.

6.3 Methods and materials for containment and cleaning up

Collect spilled or leaked chemical substances thoroughly using a vacuum cleaner or other appropriate equipment. Do not allow dust to disperse. Store the collected material in suitable, labeled containers awaiting disposal.

Dispose of the material according to the procedures described in Section 13

(Disposal Considerations).

SECTION 7: HANDLING AND STORAGE

7.1 Precautions for safe handling

Avoid all personal contact. Wash thoroughly after handling. Launder contaminated clothing prior to use. Do not eat, drink and smoke in working area. Good Industrial hygiene should be applied. Avoid inhaling dust of the product. Ensure adequate ventilation. Related equipment should be grounded and protected against static electricity buildup.



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7.2 Conditions for safe storage, including any incompatibilities

Keep container tightly closed. Store in a cool and well-ventilated place. Keep away from direct sunlight, fire and ignition sources. Keep away from strong oxidants, strong acid and PVC solvent (organic solvent) Further details on proper arrangement and storage methods can be found in the "User Manual" document.





Store in dry place

Good storage measures should be implemented for this product, and other materials should be kept separate from the area.

SECTION 8: EXPOSURE CONTROLS/PERSONAL PROTECTION

8.1 **Control parameters**

Occupational Exposure Limits

Substance	Country/organization	Occupational Exposure Limits	
Particulates Not Otherwise Regulated, Total and Respirable Dust (dust, nuisance	USA/OSHA	8-hour TWA: 15 mg/m³ (Total dust); 5 mg/m³ (Respirable)	
dust, inert dust) [4]:	USA/California OSHA	8-hour TWA: 10 mg/m³ (Nuisance dust); 5 mg/m³ (Respirable)	
	USA/ACGIH	8-hour TWA: 10 mg/m³ (Nuisance dust); 1 mg/m³ (Respirable)	
Dust, (PVC) [5]	Canada	8-hour TWA: 1 mg/m³ (Respirable)	
	Chaina	8-hour TWA: 5 mg/m³ (Respirable)	

Appropriate engineering 8.2 controls

For areas where PVC is heat-formed, a ventilation system should be installed to remove residual hydrochloric acid gas and vinyl chloride monomer vapors. Do not inhale dust or chemical vapors.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection



Wear Safety glasses or goggle.

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Skin protection



Wear dust resistant gloves and chemical or oil resistant apron and clothes.

Respiratory protection



In case of insufficient ventilation, use personal protective equipment. Recommended respirator or filter is N95 (NIOSH standard) or P2 (EN143 standard). If oil particles are also present, use a P100 filter (NIOSH standard). [6]

Thermal hazards No data are available.

SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES

9.1 Information on basic physical and chemical properties

Physical state Solid (powder)

Color White

Odor Normally odorless. If stored in unventilated packaging for

an extended period, a slight odor may develop.

Melting point/freezing point No data are available.

Initial boiling point/boiling range No data are available.

Flammability No data are available.

Lower and upper explosive limits/flammability limit Not applicable.

Flash point 391 °C

Auto-ignition temperature 450 °C

Decomposition temperature No data are available.

pH Not applicable.

Kinematic viscosity Not applicable.

Solubility Insoluble in water.

Partition coefficient: n-octanol/water No data are available.

Vapour pressure No data are available.

Density or Relative density No data are available.

Relative vapour density Not applicable.

Particle characteristics No data are available.

9.2 Other information No data are available.



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SECTION 10: STABILITY AND REACTIVITY

10.1 Reactivity No data are available.

10.2 Chemical stability Stable under normal temperature and pressure.

10.3 Possibility of hazardous No data are available.

reactions

10.4 Conditions to avoid Heat, sunlight, flame and ignition sources.

10.5 Incompatible materials Strong oxidants, strong acids and organic solvents. [13]

10.6 Hazardous decomposition Thermal decomposition will release hydrogen chloride gas and carbon

products monoxide gas. It may decompose into benzene, aromatic, and aliphatic

hydrocarbons.

SECTION 11: TOXICOLOGICAL INFORMATION

11.1 Information on toxicological effects

Acute toxicity

Ingestion No data are available.

Skin contact No data are available.

Inhalation No data are available.

Skin corrosion/irritation No data are available.

Serious eye damage/irritation No data are available.

Respiratory or skin sensitization No data are available.

Germ cell mutagenicity No data are available.

Carcinogenicity According to International Agency for Research on Cancer (IARC), PVC is

classified as Group 3 Not classifiable as to its carcinogenicity to humans. [7][8]

Reproductive toxicity No data are available.

STOT - single exposure PVC is classified as Specific target organ toxicity - single exposure, Category

3 (Respiratory tract irritation) by Japanese government. [9]

STOT - repeated exposure No data are available.

Aspiration hazard No data are available.

11.2 ข้อมูลอื่น ๆ PVC may contain trace amounts of residual vinyl chloride monomer, which is

classified as a Category 1A carcinogen under EU regulations. [10]



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SECTION 12: ECOLOGICAL

The information given is based on data available for the material, the components of the material, and similar materials.

12.1 Toxicity No data are available.

12.2 Persistence and degradability No data are available.

12.3 **Bioaccumulative potential** No data are available.

Mobility in soil 12.4 No data are available.

12.5 Other adverse effects The decomposition of PVC microplastics releases hazardous substances that

negatively impact soil ecosystems, affecting soil animals, plants, microbes,

enzyme activity, and nutrient cycling. [11]

SECTION 13: DISPOSAL CONSIDERATIONS

13.1 **Disposal methods**

Waste Disposal:

Waste generated is classified as chemical waste. Dispose of it by secure landfill or by incineration with pollution control, in compliance with local regulations. Users must consider if the material can be reused or reprocessed.

As the supplier, the company has no control over the disposal or production processes of those possessing or using the substance. The disposal methods mentioned above apply to the product in the condition specified in Section 3 (Composition/Information on Ingredients) of the Safety Data Sheet, and do not include substances mixed with other materials or contaminants.

Packaging Disposal:

PVC resin packaging is chemical-contaminated waste. It must not be reused or filled with other substances. Disposal of such packaging must comply with local laws and regulations.

SECTION 14: TRANSPORT INFORMATION

		ADR/RID	IMDG	IATA/ICAO
14.1	UN Number	Not assigned	Not assigned	Not assigned
14.2	UN proper shipping name	Not assigned	Not assigned	Not assigned
14.3	Transport hazard class	Not assigned	Not assigned	Not assigned
14.4	Packing group	Not assigned	Not assigned	Not assigned
14.5	Environmental hazards	Not assigned	Not assigned	Not assigned
14.6	Special precautions for user	No special precautions.		



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14.7 Annex II of MARPOL73/78 and the IBC Code

No data are available.

SECTION 15: REGULATORY INFORMATION

15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

OSHA 29 CFR 1910.1017: PVC resin may contain trace amounts of residual vinyl chloride monomer.

> Under normal operating conditions, which include adequate ventilation and an 8-hour OSHA work shift, detectable vinyl chloride monomer levels must not exceed PEL of 1.0 PPM, action level of 0.5 PPM, or C/STEL of 5.0 PPM.

TSCA (40 CFR 710): Polyvinylchloride is included in TSCA list.

PVC resin may contain trace amounts of residual vinyl chloride monomer, California Proposition 65:

which is classified as carcinogen under California regulations.

SECTION 16: OTHER INFORMATION

16.1 NFPA



Health Hazards 1 Fire Hazards 1 Reactivity O Specific Hazardous None

16.2 **Abbreviations** and acronyms::

ADR/RID Agreement concerning the International Carriage of Dangerous Goods by Road / Regulations

concerning the International Carriage of Dangerous Goods by Rail.

GHS Globally Harmonized System of Classification and Labelling of Chemicals.

IATA/ICAO International Air Transport Association / International Civil Aviation Organization.

IMDG International Maritime Dangerous Goods.

OSHA Occupational Safety and Health Administration

TWA Time Weighted Average

16.3 References: [1] Review of pulmonary effects of poly (vinyl chloride) and vinyl chloride exposure.

Environmental Health Perspectives, Volume 41, Pages 167 - 169

https://doi.org/10.1289/ehp.8141167

[2] Vinyl Chloride and Polyvinyl Chloride Exposure and Occupational Lung Disease, CHEST,

Volume 78, Issue 6, 826 - 828 https://doi.org/10.1378/chest.78.6.826



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[3] Anastasiia O. Nosova, Mayya V. Uspenskaya, Ecotoxicological effects and detection features of polyvinyl chloride microplastics in soils: A review, Environmental Advances, Volume 13, 2023, 100437, ISSN 2666-7657, https://doi.org/10.1016/j.envadv.2023.100437.

[4] US Department of Labor, Occupational Safety and Health Administration; https://www.osha.gov/chemicaldata/801

[5] GESTIS - International Limit Values for Chemical Agents (GESTIS-ILV); https://ilv.ifa.dguv.de/limitvalues/25415

- [6] 3M Respirator selection guide.
- [7] International Agency for Research on Cancer (IARC), Monographs Volume 19: (1979) Some Monomers, Plastics and Synthetic Elastomers, and Acrolein.
- [8] International Agency for Research on Cancer (IARC), Monographs Volume Sup 7: Overall Evaluations of Carcinogenicity: An Updating of IARC Monographs Volumes 1 to 42, 1987; 440 pages; ISBN 92-832-1411-0 (out of print)
- [9] National Institute of Technology and Evaluation, NITE-CHRIP; https://www.cheminfo.nite.go.jp/chem/english/ghs/15-mhlw-0019e.html
- [10] Harmonised classification Annex VI of Regulation (EC) No 1272/2008 (CLP Regulation) https://echa.europa.eu/information-on-chemicals/cl-inventory-database/-/discli/details/11609
- [11] Zhang M, Buekens A, Jiang X, Li X. Dioxins and polyvinylchloride in combustion and fires. Waste Manag Res. 2015 Jul;33(7):630-43. doi: 10.1177/0734242X15590651. PMID: 26185164.
- [12] International Chemical Safety Cards (ICSC) of POLYVINYL CHLORIDE;
- $https://chemicalsafety.ilo.org/dyn/icsc/showcard.display?p_lang=en\&p_card_id=1487\&p_version=2$

[13] Edo, G.I., Ndudi, W., Ali, A.B.M. et al. Poly(vinyl chloride) (PVC): an updated review of its properties, polymerization, modification, recycling, and applications. J Mater Sci 59, 21605–21648 (2024). https://doi.org/10.1007/s10853-024-10471-4

Disclaimers

The types of applications specified here are for reference only. Customers are responsible for verifying and testing the product's suitability and application to ensure it meets their specific purposes and uses. Customers are solely responsible for the product's appropriateness, safety, lawful use, and handling processes.

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